

# Metabolism

CE421/521

Lecture Notes

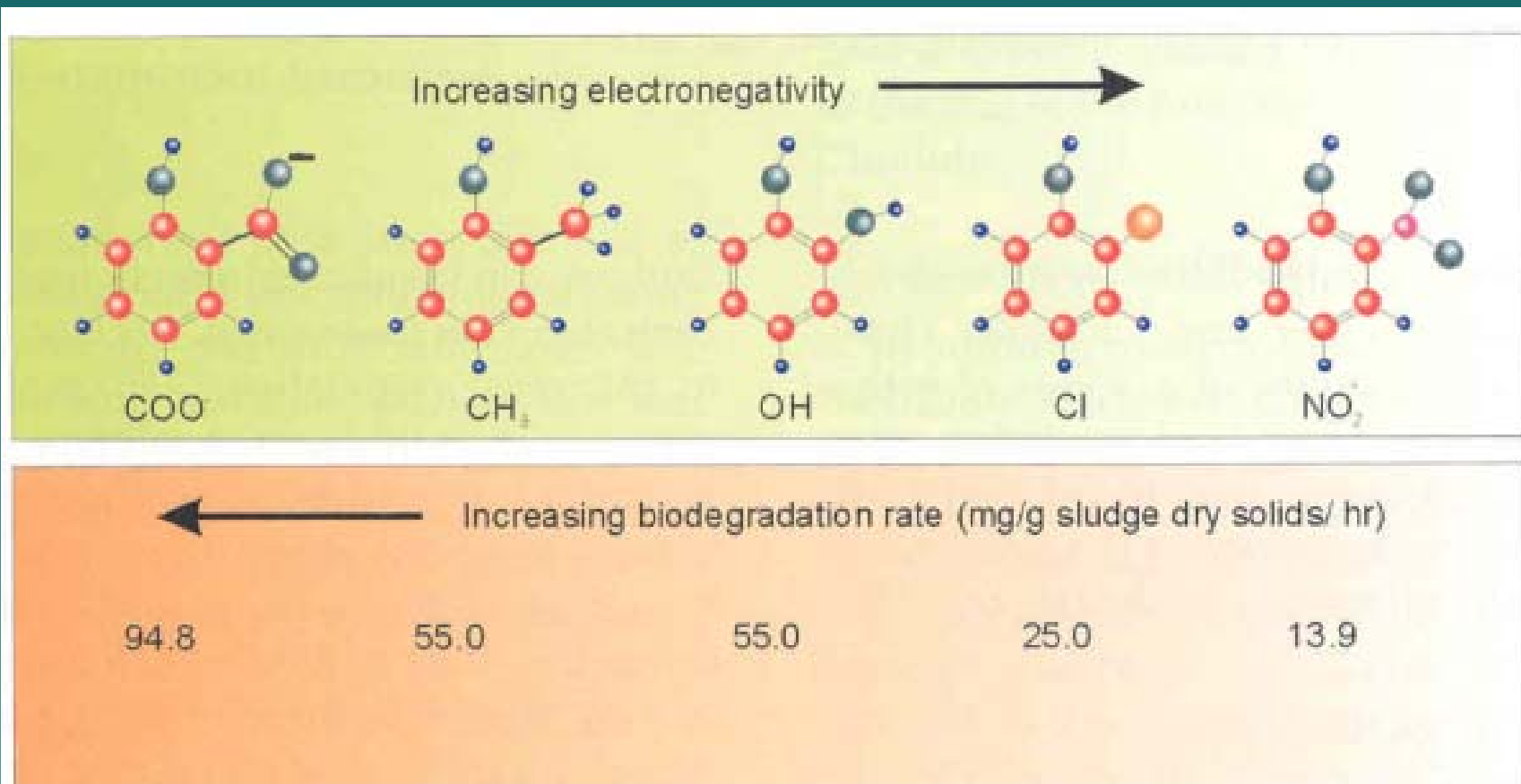
Sept. 12, 2006

# Biodegradation Processes

- ◆ cometabolism, mineralization, and biodegradation
  - biodegradation: b\_\_\_\_\_ of organic compounds by microorganisms
  - mineralization: c\_\_\_\_\_ biodegradation of organic compounds to CO<sub>2</sub> and water
  - cometabolism: breakdown of an organic compound where the degrading community derives no b\_\_\_\_\_ (i.e., carbon or energy) from degradation (requires a growth substrate) TCE degradation is a common example of cometabolism via methane monooxygenase

# Structure, Toxicity, and Biodegradability

- ◆ Factors determining the rate and potential for biodegradation:
  - G\_\_\_\_\_ potential. Appropriate genes for transport and metabolism of substrate
  - B\_\_\_\_\_. Limited water solubility may limit biodegradation.
  - Contaminant s\_\_\_\_\_: steric and electronic effects.
    - ◆ Steric effects include substituent groups h\_\_\_\_\_ recognition of active site for enzyme attachment and activity.
    - ◆ Electronic effects include the extent to which the substituent group e\_\_\_\_\_ interferes with the interaction between the enzyme active site and the contaminant
  - T\_\_\_\_\_ or inhibitory effect of the contaminant on cellular metabolism



**FIGURE 16.11** Various ortho-substituted phenols and their respective biodegradation rates. (Adapted from Pitter and Chudoba, 1990.)

# Ten Growth Requirements for Microorganisms

1. \_\_\_\_\_ source
2. \_\_\_\_\_ source
3. Terminal \_\_\_\_\_ acceptor
4. \_\_\_\_\_ nutrients: C, N, H, O, P, K, S
5. \_\_\_\_\_ nutrients: Fe, Ni, Co, Mb, Zn, etc.
6. M \_\_\_\_\_
7. Appropriate t \_\_\_\_\_
8. Appropriate p \_\_\_\_\_
9. Absence of I \_\_\_\_\_
10. Mixing/c \_\_\_\_\_

# Metabolism

- ◆ catabolism – produce energy to drive cell machinery, exergonic
- ◆ anabolism – biosynthetic reactions, endogonic
- ◆ Energy storage:
  - ADP – adenosine diphosphate
  - ATP – adenosine triphosphate7500 cal/bond

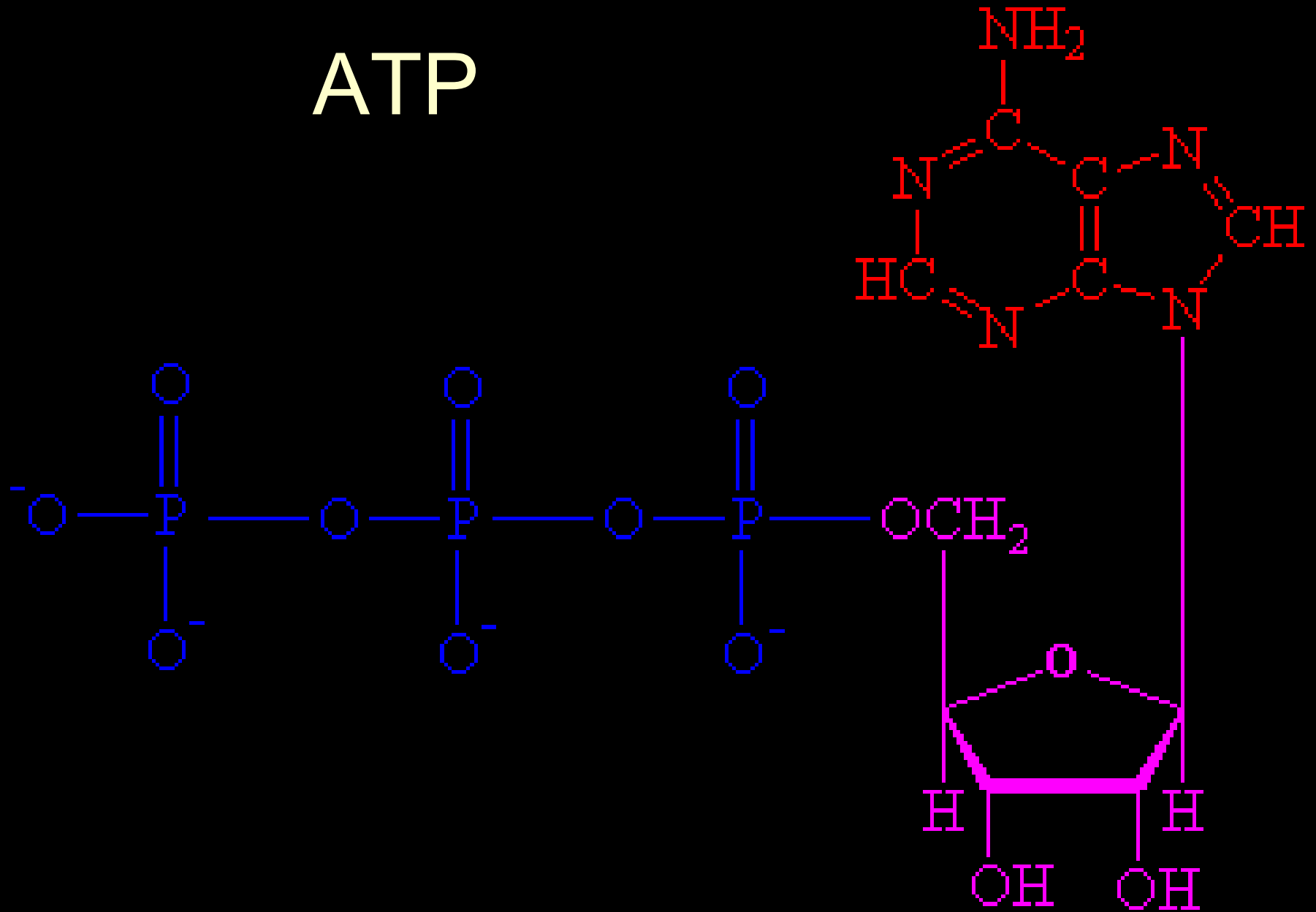


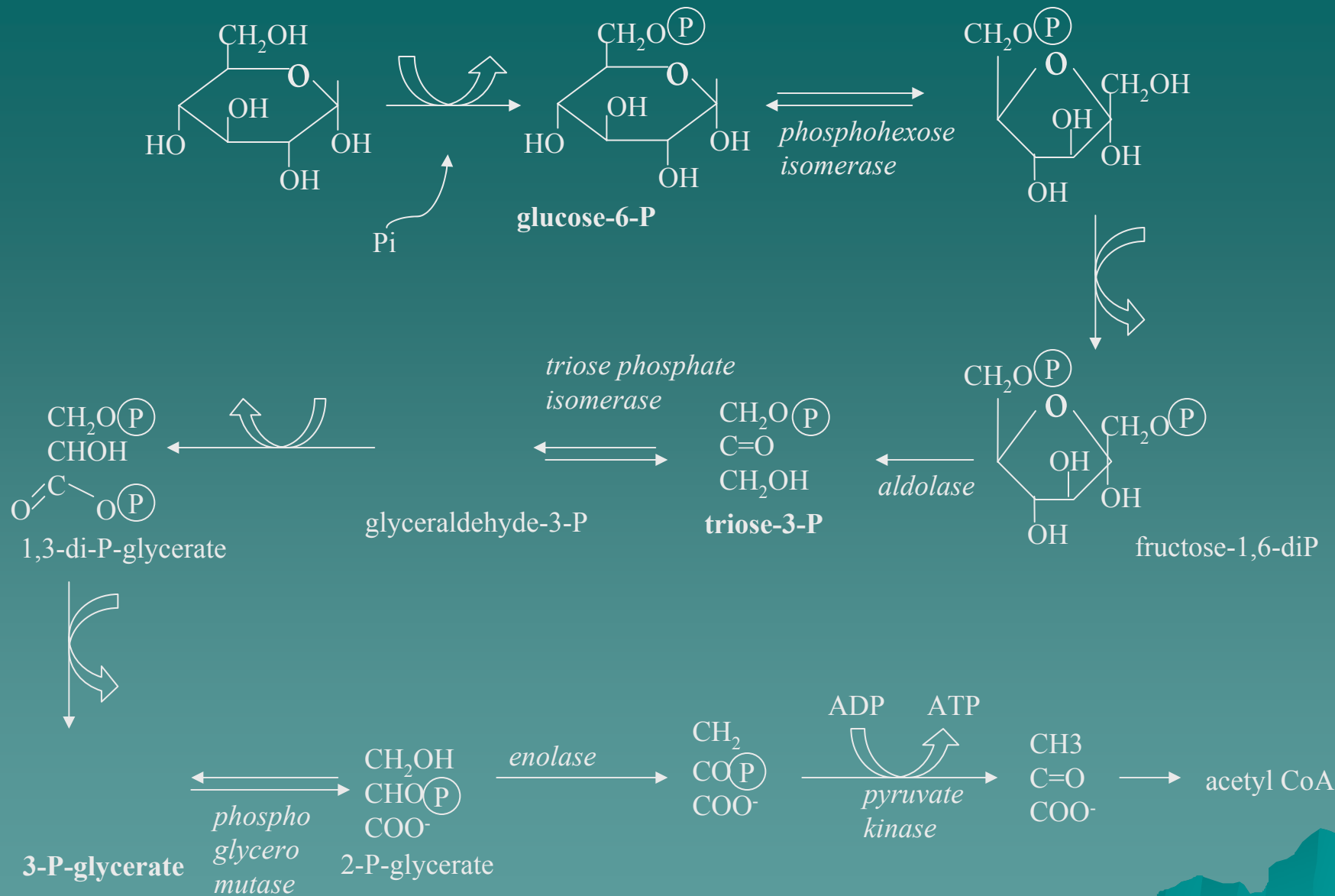
# ATP Production

- ◆ formed in 3 ways:
  1. Substrate level phosphorylation (SLP): occurs during fermentation, e.g., glycolysis - breakdown of glucose in EMP pathway to pyruvate produces 4 ATP, consumes 2 ATP, nets 2 ATP
  2. Oxidative phosphorylation: electron transport, proton motive force
  3. Photophosphorylation



# ATP





# Embden-Meyerhof-Parnas

# Aerobic Degradation of Glucose:



# Electron Transport System

- ◆ at the end of the EMP, Krebs (TCA) cycle, Pentose phosphate, or Entner-Doudoroff pathway:
  - reduced nucleotides (electron carriers), some ATP from substrate level phosphorylation
  - reduced nucleotides enter the electron transport system in cytoplasmic membrane

# Steps in ETS:

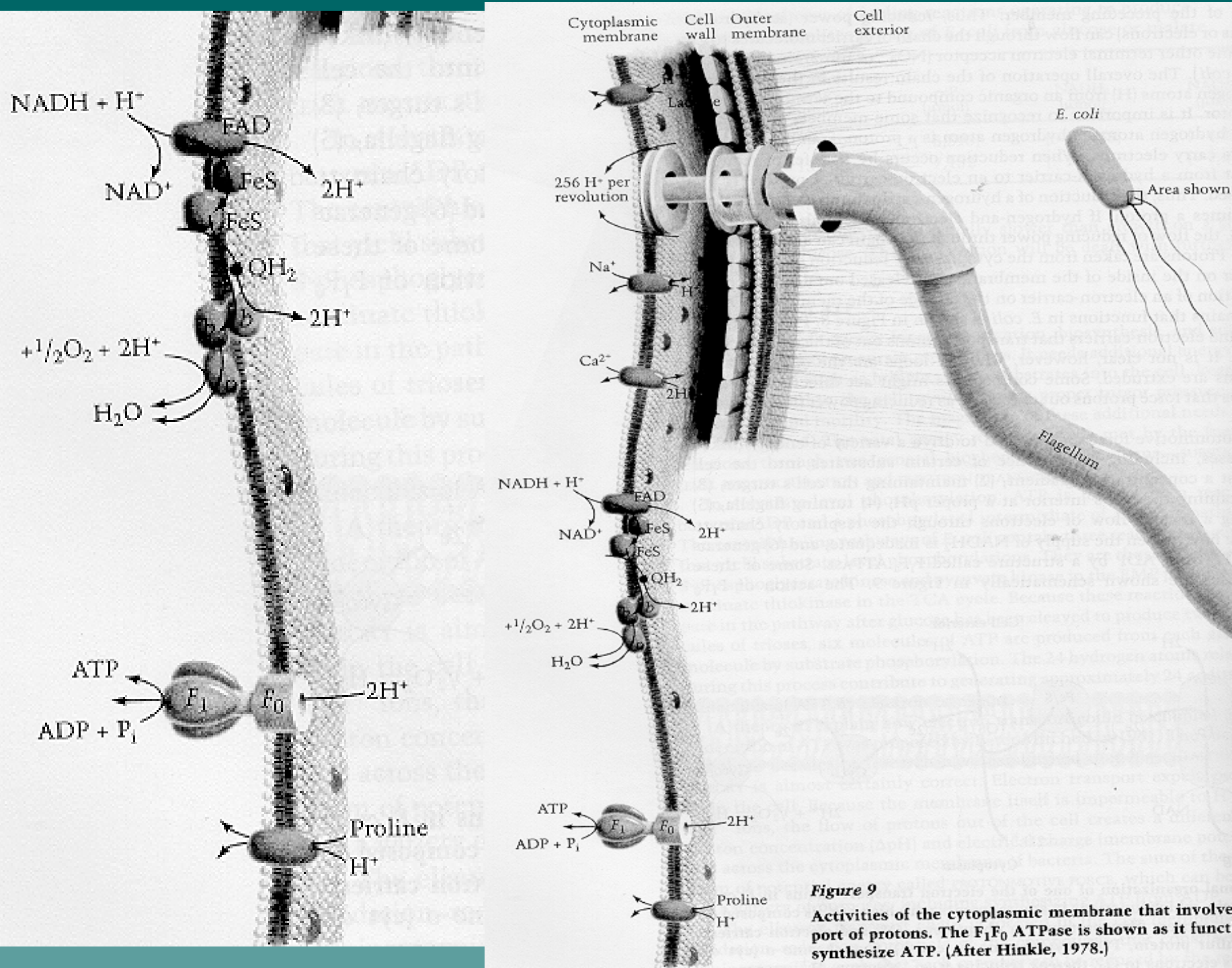
1. Electrons from substrate transferred to reduced carriers ( $\text{NAD} \rightarrow \text{NADH} + \text{H}^+$ ,  $\text{NADP} \rightarrow \text{NADPH} + \text{H}^+$ )
2. Flavoproteins accept  $\text{H}_2$  & transport it outside the cell
3. electrons are transported to the iron/sulfur protein
4. Quinones accept electrons and transport additional  $\text{H}^+$  outside the cell

# ETS on Membrane



# ATP Production from H gradient





**Figure 9**

Activities of the cytoplasmic membrane that involve transport of protons. The F<sub>1</sub>F<sub>0</sub> ATPase is shown as it functions to synthesize ATP. (After Hinkle, 1978.)